

Engine Emission & Controls

The emissions are

CO₂

CO

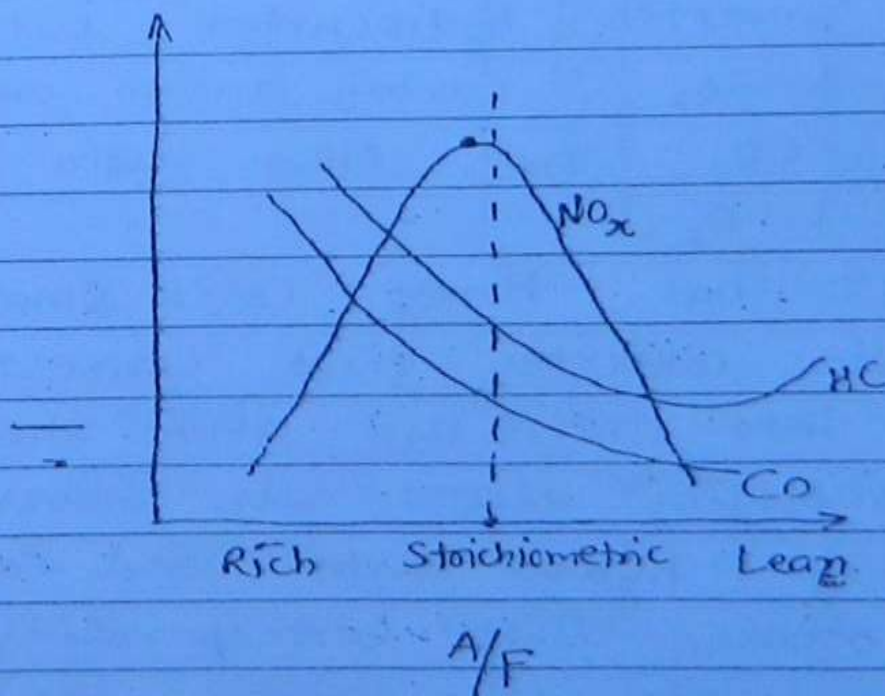
Unburnt HC

Oxides of Nitrogen (NO_x)

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Of these Carbon mono oxide, hydrocarbons & oxides of Nitrogens are main pollutants.

Oxide of N₂ are formed when the temp.s are very-very high and when there is sufficient availability of oxygen.



Emission Control :-

Change in engine design so as to keep lower compression ratios.

Slightly leaner mixture must be provided.

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An addition of after burner to the exhaust can completely burn partially burnt Hydro Carbons in exhaust gases.

Catalytic Converter consists of Platinum, Palladium & Rhodium and it converts hydrocarbon into H_2O & CO_2 , Carbon mono oxide into CO_2 and NO_x into N_2 & O_2 .

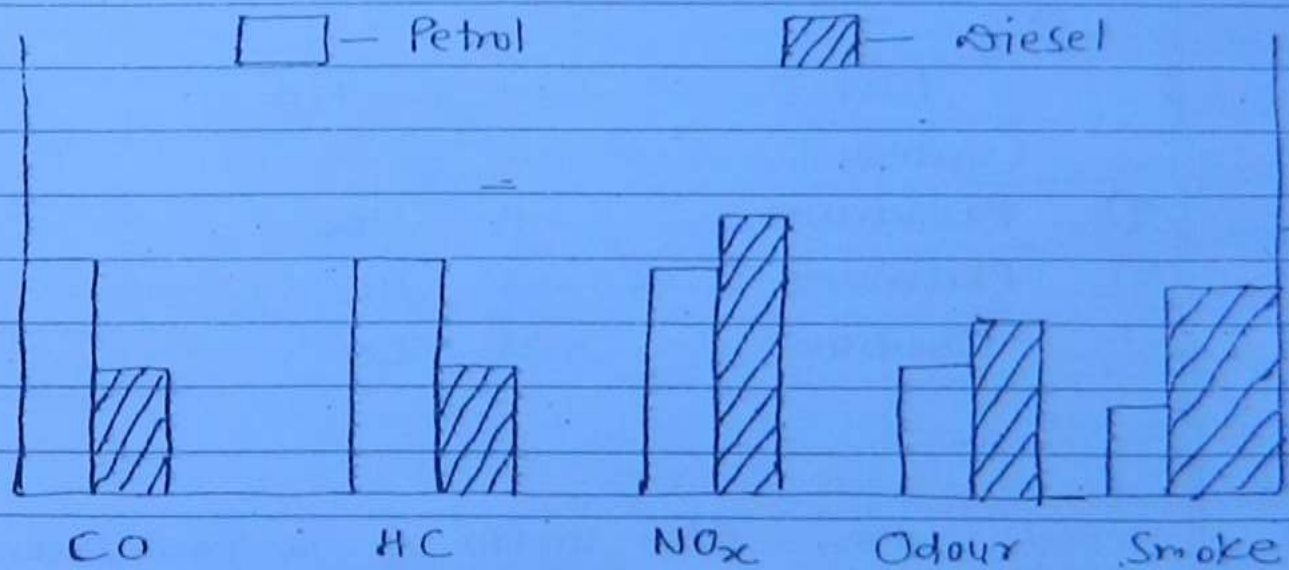
It has Honey Comb Structure.

Catalytic converter first converts NO_x into N_2 & O_2 and then by supplying excess air conversion of Hydro Carbon and Carbon monoxide into corresponding oxides takes place.

Drawbacks of Catalytic Converter :-

(12)

- * The exhaust temp. are more when catalytic converter is fitted.
- * If lead is used in fuel this lead attacks the converter and hence decreases conversion efficiency.
- * If sulphur is present in fuel catalytic converter converts sulphur into oxides of sulphur. And this may result in acid rain.



Diesel engines if properly maintain have very little CO in their exhaust. And a small amount of smoke. While petrol engine exhaust significant amount of CO and unburnt HC. So, diesel engine is cleaner as compare to petrol engine and pollute less.

However this fact is not generally noted because a small increase in smoke becomes highly visible. while CO and Unburnt HC can not be noticed so easily.

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match the following

List-I

List-II

(Catalyst)

- | | |
|--------------|--------------------|
| A) Palladium | 1) NO _x |
| B) Platinum | 2) HC |
| C) Rhodium | 3) CO |

Find the A/F ratio of a Four stroke single cylinder engine with fuel consumption time for 10 cc is 20.4 sec. and air consumption time for 0.1 m³ is 16.3 sec.

Assume density of air as 1.175 kg/m³ and specific gravity of fuel 0.7

Solⁿ:

$$V_f = 10 \text{ cc} / 20.4 \text{ sec.}$$

$$V_a = 0.1 \text{ m}^3 / 16.3 \text{ sec.}$$

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$$\rho_a = m_a / V_a$$

$$m_a = \rho_a V_a$$

$$= 1.175 \frac{\text{kg}}{\text{m}^3} \times \frac{0.1 \text{ m}^3}{16.3 \text{ sec.}}$$

$$m_a = 7.2 \times 10^{-3} \text{ kg/s}$$

$$\rho_f = \frac{m_f}{V_f}$$

$$m_f = \rho_f \times V_f$$

$$= \frac{700 \times 10 \times 10^{-6}}{20.4}$$

$$m_f = 0.34 \times 10^{-3}$$

$$\therefore \frac{A}{F} = \frac{m_a}{m_f}$$

$$= \frac{7.2 \times 10^{-3}}{0.34 \times 10^{-3}}$$

$$\frac{A}{F} = 21$$

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③

An 8-cylinder, four stroke engine of 9 cm bore and 8 cm stroke with a compression ratio of 7 is tested at 4500 rpm on a dynamometer which has 54 cm radius. During a 10 min. ^{beam} test the dynamometer scale reading was 42 kg and the engine consume 4.4 kg of fuel having a calorific value of 44000 kJ/kg. Air at 27°C and 1 bar was supplied to carburetor at the rate of 6 kg/min. Find

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- i. Brake power
- ii. Brake sp. fuel consumption
- iii. Brake sp. Air consumption
- iv. Brake thermal effⁿ.
- v. Volumetric effⁿ.
- vi. A/F ratio

Soln: -

4-stroke
 $K = 8$
 $D = 9 \text{ cm}$
 $L = 8 \text{ cm}$
 $r = 7$
 $N = 4500 \text{ rpm}$
 Dyna. radius = 54 cm
 $t = 10 \text{ min}$

$$m = 42 \text{ kg}$$

$$m_f = 4.4 \text{ kg}$$

$$m_f = \frac{4.4}{10} \text{ kg/sec}$$

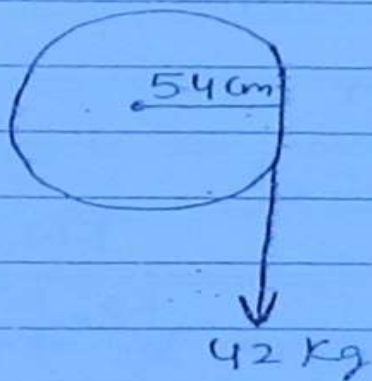
$$CV = 44000 \text{ kJ/kg}$$

$$T_1 = 300\text{K}$$

$$P_1 = 1 \text{ bar}$$

$$m_a = 6 \text{ kg/min}$$

125



$$T = F \times r$$

$$= 42 \times 9.81 \times 0.54$$

$$= 222.49$$

$$F = 42 \times 9.81$$

$$BP = \frac{2\pi NT}{60}$$

$$= \frac{2\pi \times 4500 \times 222.49}{60}$$

$$= 104.8 \times 10^3 \text{ W}$$

(2)

$$BP = 104.8 \text{ kW}$$

BSFC

$$BSFC = \frac{m_f}{B.P.}$$

m_f in kg/hr.

$$= \frac{4.4 \times 60}{104.8 \times 10} = 0.251$$

$$m_f = \frac{4.4 \times 60}{10}$$

$$= 0.251 \text{ kg/kwhr}$$

(iii)

$$\begin{aligned} \text{BSAC} &= \frac{m_a}{\text{B.P.}} \\ &= \frac{6 \times 60}{104.8} \quad (126) \\ &= 0.0572 \times 60 \\ \text{BSAC} &= 3.435 \text{ Kg/kwhr.} \end{aligned}$$

(iv)

Brake Thermal effⁿ.

$$\begin{aligned} \eta_{\text{bth}} &= \frac{\text{B.P.}}{m_f \times \text{C.V.}} \quad \frac{\text{Kw}}{\frac{\text{Kg} \times \text{KJ}}{\text{Sec.} \times \text{Kg}}} \\ &= \frac{104.8 \text{ Kw}}{4.4 \frac{\text{Kg}}{\text{Sec.}} \times 44000 \frac{\text{KJ}}{\text{Kg}}} \end{aligned}$$

$$\eta_{\text{bth}} = 0.3248$$

(v)

$$\eta_{\text{bth}} = 32.48 \%$$

(v)

$$\frac{A}{F} = \frac{m_a}{m_f}$$

$$= \frac{6 \text{ Kg/min.}}{4.4 \text{ Kg/min.}}$$

$$\frac{A}{F} = 13.63$$

(127)

(vi) Volumetric efficiency.

$$\eta_v = \frac{\text{Actual volume of Air/cycle} = \frac{V_a}{V_s}}{\text{Swept volume}}$$

$$\rho_a = \frac{m_a}{V_a}$$

$$V_a = \frac{m_a}{\rho_a}$$

$$P = PRT$$

$$\rho_a = \frac{\rho_a}{R_a T_a} = \frac{100}{0.287 \times 300}$$

$$\rho_a = 1.1614 \text{ Kg/m}^3$$

$$V_a = \frac{6 \text{ Kg/min}}{1.1614 \text{ Kg/m}^3}$$

$$V_a = 5.166 \text{ m}^3/\text{min}$$

For 4-stroke engine

$$4500 \text{ rpm} = \frac{4500}{2} = 2250 \text{ cycle/min}$$

$$V_a / \text{cycle} = \frac{5.166 \text{ m}^3/\text{min}}{2250 \text{ cycle}/\text{min}}$$

$$= 2.296 \times 10^{-3} \text{ m}^3/\text{cycle}$$

$$V_s = \frac{\pi}{4} D^2 L K \quad (128)$$

$$= \frac{\pi}{4} (0.09)^2 (0.08) 8 \text{ m}^3$$

$$= 4.07 \times 10^{-3} \text{ m}^3$$

$$\eta_{\text{vol.}} = \frac{V_a}{V_s}$$

$$= \frac{2.296 \times 10^{-3}}{4.07 \times 10^{-3}}$$

$$\eta_{\text{vol.}} = 0.5639$$

(or)

$$\eta_{\text{vol.}} = 56.39 \%$$

Note: The volumetric effⁿ. of above engine is very low and volumetric effⁿ. of a well design engine is in the range of 75 - 85 %

(129)

Gate
(4)

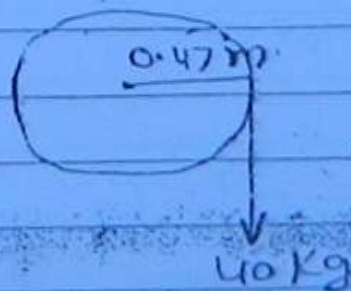
The following details were noted in a test on a 4-cylinder, 4-stroke engine

| | |
|---------------------|-----------------|
| Bore | 100 mm |
| Stroke | 120 mm |
| Speed | 1600 rpm |
| Fuel Consumption | 0.2 Kg/min |
| C.V. of fuel | 44000 KJ/Kg |
| Brake load | 40 Kg |
| Brake circumference | 300 cm (or) 3 m |
| η_m | 80 % |

Calculate

- (i) Brake thermal effⁿ.
- (ii) Indicated thermal effⁿ.
- (iii) Indicated mean effective pressure &
- (iv) Brake specific Fuel Consumption.

Solⁿ:-



$$F = 9.81 \times 40$$

$$= 392.4$$

$$\text{Circumference} = 2\pi r$$

$$3 = 2\pi r$$

$$r = 0.477 \text{ m}$$

$$T = F \times r$$

$$= 392.4 \times 0.477$$

$$= 187.35$$

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$$\text{Brake power} = \frac{2\pi NT}{60}$$

$$= \frac{2\pi \times 1600 \times 187.35}{60}$$

$$\text{BP} = 31392 \text{ W}$$

(22)

$$\text{BP} = 31.39 \text{ kW}$$

i) Brake thermal effⁿ.

$$\eta_{\text{bth}} = \frac{\text{BP}}{m_f \times C \cdot V}$$

$$= \frac{31.39 \text{ kW}}{0.2 \frac{\text{kg}}{60 \text{ sec}} \times 46000 \frac{\text{KJ}}{\text{kg}}}$$

$$= \frac{31.39 \text{ kW}}{1.5333 \text{ kW}}$$

$$\eta_{\text{bth}} = 0.2156$$

(23)

$$\eta_{\text{bth}} = 21.56\%$$

bth

(ii) Indicated Thermal effⁿ.

$$\eta_{ith} = \frac{\eta_{bth}}{\eta_m}$$

(131)

$$= \frac{0.2156}{0.8}$$

$$\eta_{ith} = 0.269$$

(or)

$$\eta_{ith} = 26.9\%$$

(iii) Indicated mean effective pressure

$$IP = \frac{P_m IANK}{2 \times 60}$$

$$\eta_m = \frac{BP}{I.P.}$$

$$0.8 = \frac{31.36}{I.P.}$$

$$I.P. = 39.24 \text{ kW}$$

$$39.24 = \frac{P_m \times (0.12) \times \frac{\pi}{4} (0.1)^2 \times 1800 \times 4}{60 \times 2}$$

$$P_m = 780.65 \text{ kPa}$$

(or)

$$P_m = 780.65 \times 10^3 \text{ N/m}^2 \quad (or) \quad P_m = 7.8065 \text{ bar}$$

(iv)

Brake Specific Fuel Consumption

$$BSFC = \frac{m_f}{B.P.}$$

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$$= \frac{0.2 \times 60 \text{ Kg}}{31.39 \text{ Kw hr.}}$$

$$BSFC = 0.3826 \text{ Kg/Kw hr.}$$

(v) The power output of a 6-cylinder, 4-stroke engine by measure means of a dynamometer for which the law is

$$b_p = \frac{WN}{20000} \text{ Kw}$$

where

$W =$ brake load in N

$N =$ speed in rpm

The following data may be used

| | |
|-----------------|----------|
| Throat diameter | 30 mm |
| Bore | 100 mm |
| Stroke | 120 mm |
| Brake load | 500 N |
| Speed | 2000 rpm |

$\frac{C}{H}$ by mass

(1)

$\frac{83}{17}$

(183)

Coeff. for discharge for venturi

0.6

atmp. pressure

1 bar

pressure drop across venturi

14.5 cm of H₂O

Time taken for 100cc fuel

Consumption

20 sec.

atmp. Temp.

27°C

Fuel density

831 kg/m³

calculate

(i) Brake power

(ii) Brake Torque

(iii) BSFC

(iv) Volumetric eff.

(v) Percentage excess of air

Soln:- * Assume Air as incompressible

(i) Brake power

$$bP = \frac{WN}{20000}$$

$$bP = \frac{560 \times 2400}{20000}$$

$$bP = 67.2 \text{ kW}$$

(ii) Brake Torque

$$bP = \frac{2 \times NT}{60} \text{ kW}$$

$$67.2 \times 10^3 = \frac{2\pi \times 2400 \times T}{60}$$

(134)

$$T = \frac{67.2 \times 10^3 \times 60}{2\pi \times 2400}$$

$$T = 267.3 \text{ N-m}$$

(iii) % excess of Air

$$\left(\frac{A}{F}\right)_{st.} = \frac{100}{23} \left(\frac{8}{3} C + 8H_2\right)$$

$$\left(\frac{A}{F}\right)_{st.} = \frac{100}{23} \left[\frac{8}{3} \times 0.83 + 8 \times (0.17)\right]$$

$$\left(\frac{A}{F}\right)_{st.} = 15.536$$

Actual Air/Fuel ratio

$$V_f = 100 \text{ cm}^3 / 20 \text{ sec}$$

$$\rho_f = 831 \text{ kg/m}^3$$

$$P = \frac{m_f}{V_f}$$

$$V_f = 15.536$$

$$m_f = \rho_f \times V_f$$

$$= 831 \frac{\text{kg}}{\text{m}^3} \times \frac{100 \times 10^{-6} \text{ m}^3}{20 \text{ sec}}$$

(135)

$$m_f = 4.155 \times 10^{-3} \text{ kg/sec.}$$

$$m_a = C_d A_2 \sqrt{2 \rho_a (P_1 - P_2)}$$

$$(P_1 - P_2) = 14.5 \text{ cm. of Hg}$$

$$P = \rho R T$$

$$\rho_a = \frac{P}{R T}$$

$$P = \rho g h$$

$$\rho_a = \frac{100}{0.287 \times 300}$$

$$(P_1 - P_2) = 13.6 \times 10^3 \times 9.81 \times \frac{14.5}{100}$$

$$\rho_a = 1.16$$

$$P_1 - P_2 = 19345.3 \text{ N/m}^2$$

\therefore

$$m_a = 0.6 \times \frac{\pi}{4} (0.03)^2 \times \sqrt{2 \times 1.16 (19345.3)}$$

$$m_a = 0.089 \text{ kg/sec.}$$

$$\left(\frac{A}{F}\right)_{\text{act.}} = \frac{0.089}{4.155 \times 10^{-3}}$$

$$= 21.62$$

$$\% \text{ Excess} = \frac{\left(\frac{A}{F}\right)_{\text{act.}} - \left(\frac{A}{F}\right)_{\text{st.}}}{\left(\frac{A}{F}\right)_{\text{st.}}}$$

$$\textcircled{136}$$

$$\% \text{ excess} = \frac{21.62 - 15.536}{15.536}$$

$$\% \text{ excess} = 0.391$$

or

$$\% \text{ excess} = 39.1 \%$$

iv) Brake specific fuel consumption (BSFC)

$$\text{BSFC} = \frac{m_f}{bP}$$

$$\text{BSFC} = \frac{4.155 \times 10^{-3} \times 3600}{67.2}$$

$$\text{BSFC} = 0.222 \text{ Kg/kwhr.}$$

v) Volumetric effⁿ.

$$\eta_v = \frac{V_a / \text{cycle}}{V_s}$$

$$\rho_a = \frac{m_a}{V_a}$$

(13%)

$$V_a = \frac{m_a}{\rho_a}$$

$$V_a = \frac{0.089 \text{ Kg/sec.}}{1.16 \text{ Kg/m}^3}$$

$$V_a = 0.0766 \text{ m}^3/\text{sec.}$$

For 4-st engine 2400 rot./min.

$$= \frac{2400}{2} = 1200 \text{ cycle/min.}$$

$$\frac{V_a}{\text{cycle}} = \frac{0.0766 \text{ m}^3/\text{sec.}}{\frac{1200}{60} \text{ cycle/sec.}}$$

$$\frac{V_a}{\text{cycle}} = \frac{60 \times 0.0766 \text{ m}^3}{1200}$$

$$V_a/\text{cycle} = 3.8315 \times 10^{-3} \text{ m}^3$$

$$V_s = \frac{\pi}{4} D^2 L K = \frac{\pi}{4} (0.1)^2 (0.12) \times 6$$

$$V_s = 5.652 \times 10^{-3} \text{ m}^3$$

$$\eta_v = \frac{V_a}{V_s}$$

(138)

$$= \frac{3.8315 \times 10^{-3}}{5.652 \times 10^{-3}}$$

$$\eta_v = 0.6877$$

(28)

$$\eta_v = 68.77\%$$

⑤ A 6-cylinder, 6-stroke engine delivers a brake power of 120 kW at 1600 rpm. The fuel to be used has a calorific value of 43000 kJ/kg. And its percentage composition by mass is C = 86%, H = 13% & 1% non-combustibles. The volumetric effⁿ. assume to be 80%. Indicated thermal effⁿ. is 40%. mechanical effⁿ. 80%. The air consumption is 110 excess with resp. to stoichiometric conditions.

(i) Estimate volumetric composition of dry exhaust gases.

(ii) Also determine bore & stroke of

the engine. Taking Stroke to Bore ratio $L/D = 1.5$.

Assume density of air as $1.298 \frac{kg}{m^3}$

and air contains 23 Kg of oxygen in 100 Kg.

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Soln.

4-Stroke

$$K = 4$$

$$N = 1600 \text{ rpm}$$

$$bP = 120 \text{ kW}$$

$$CV = 43000 \text{ KJ/kg}$$

$$\%C = 0.86, \quad \%H_2 = 0.13$$

$$\eta_{vol.} = 0.8$$

$$\eta_{ith.} = 0.4$$

$$\eta_m = 0.8$$

$$\left(\frac{A}{F}\right) = 110\% \text{ Excess}$$

$$\rho_a = 1.298 \text{ Kg/m}^3$$

$$\frac{L}{D} = 1.5$$

Stoichiometric A/F ratio

$$\left(\frac{A}{F}\right)_{st.} = \frac{100}{23} \left[\frac{8}{3} (0.86) + 8 (0.13) \right]$$

$$\left(\frac{A}{F}\right)_{st.} = 14.5$$

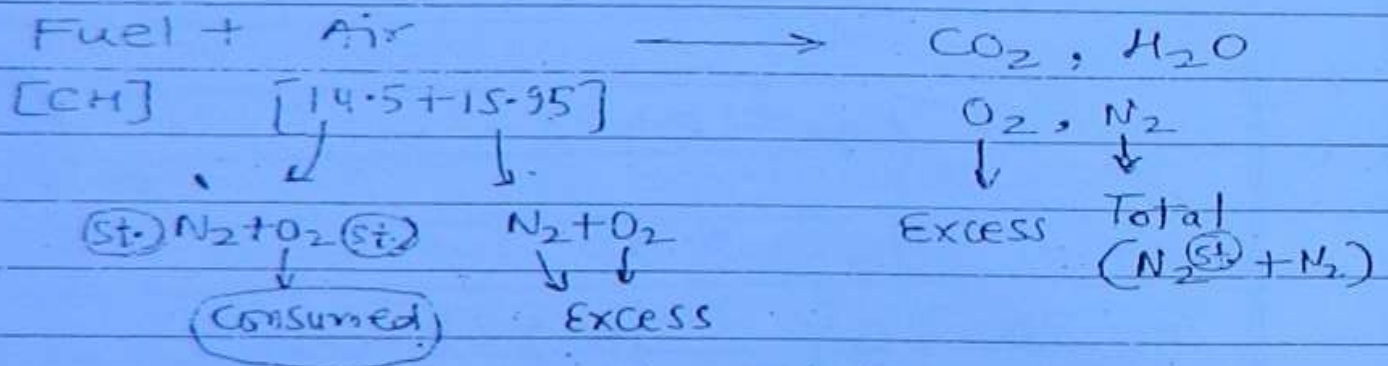
$$\left(\frac{A}{F}\right)_{act.} = 14.5 + 1.1(14.5)$$

$$\left(\frac{A}{F}\right)_{act.} = 30.45 \quad (140)$$

Excess air supply per kg fuel

$$= 30.45 - 14.5$$

$$= 15.95$$



As excess air is used stoichiometric oxygen will react with fuel resulting in CO₂ & H₂O whereas excess O₂ will come out in exhaust gases. as N₂ is inert it does not undergo any reaction and hence stoichiometric N₂ & excess N₂ will come out in exhaust gases.

$$\eta_m = \frac{b.p.}{i.p.}$$

$$i.p. = \frac{b.p.}{\eta_m} \quad (141)$$

$$= \frac{1.20 \text{ kW}}{0.8}$$

$$i.p. = 150 \text{ kW}$$

$$\eta_{ith} = \frac{i.p.}{m_f \times C.V.}$$

$$0.4 = \frac{150 \text{ kW}}{m_f \times 43000 \frac{\text{kJ}}{\text{kg}}}$$

$$m_f = 8.72 \times 10^{-3} \text{ kg/sec.}$$

For actual

$$\frac{m_a}{m_f} = 30.45$$

$$m_a = 30.45 \times m_f$$

$$m_a = 30.45 \times 8.72 \times 10^{-3}$$

$$m_a = 0.2655 \text{ kg/sec.}$$

$$\eta_{vol} = \frac{V_a / \text{cycle}}{V_s}$$

(142)

$$\rho_a = \frac{m_a}{V_a}$$

$$V_a = \frac{m_a}{\rho_a} = \frac{0.2655 \text{ kg/sec}}{1.298 \text{ kg/m}^3}$$

$$V_a = 0.2045 \text{ m}^3/\text{sec}$$

For 4-stroke engine

$$1600 \text{ rpm} \Rightarrow 800 \text{ cycle/min}$$

$$= \frac{800 \text{ cycle}}{60 \text{ sec}}$$

$$\frac{V_a}{\text{cycle}} = \frac{0.2045 \text{ m}^3/\text{sec}}{\frac{800 \text{ cycle}}{60}}$$

$$\frac{V_a}{\text{cycle}} = 0.0153 \text{ m}^3$$

$$0.8 = \frac{0.0153 \text{ m}^3}{\frac{\pi}{4} D^2 L K \text{ m}^3}$$

$$\frac{\pi D^2 L K}{4} = \frac{0.0153}{0.8}$$

$$\frac{\pi}{4} D^2 (1.5D) \times 6 = \frac{0.0153}{0.8}$$

$$D = 0.139 \text{ m.}$$

(143)

(09)

$$D = 13.9 \text{ cm.} \approx 14 \text{ cm.}$$

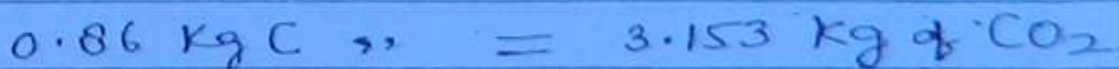
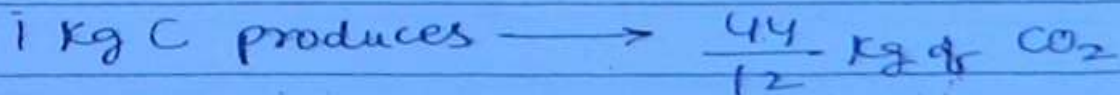
$$L = 1.5D$$

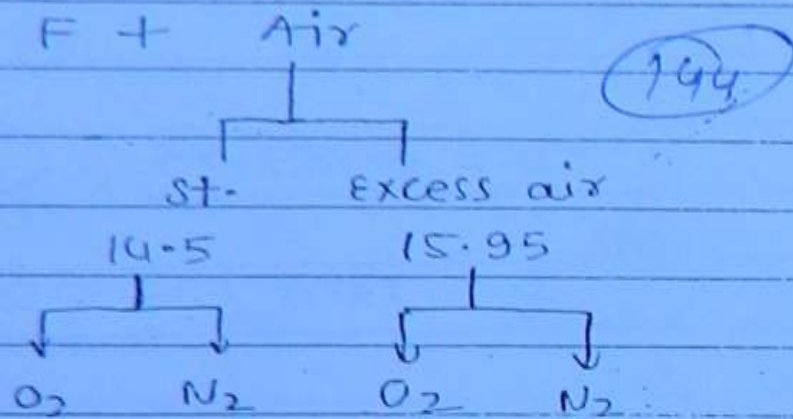
$$= 1.5 \times 14$$

$$L = 21 \text{ cm.}$$

Exhaust Gases,

$\text{CO}_2, \text{N}_2, \text{O}_2$





← 100 Kg of air — 23 Kg of O₂

15.95 Kg " — $\frac{15.95 \times 23}{100}$

15.95 Kg Air = 3.6685 Kg of O₂

← 100 Kg air — 77 Kg of N₂

30.45 — " → $\frac{77 \times 30.45}{100}$

30.45 Kg Air = 23.446 Kg of N₂

The above analysis is known as Gravimetric (or) Mass analysis

CO₂ — 3.153

O₂ — 3.668

N₂ — 23.446

Volumetric Analysis of Exhaust Gases:-

| | Grav. | Volum. |
|-----------------|--------|----------------------|
| CO ₂ | 3.153 | $3.153/44 = 0.7165$ |
| O ₂ | 3.668 | $3.668/32 = 0.1145$ |
| N ₂ | 23.446 | $23.446/28 = 0.8373$ |
| | | 1.023 |

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$$\frac{0.07165}{1.023} \times 100 = 7\%$$

$$\frac{0.1145}{1.023} \times 100 = 11.2\%$$

$$\frac{0.8373}{1.023} \times 100 = 81.8\%$$

Volumetric Analysis of Exhaust Gases :-

| | Grav. | Volam. |
|-----------------|--------|----------------------|
| CO ₂ | 3.153 | $3.153/44 = 0.7165$ |
| O ₂ | 3.668 | $3.668/32 = 0.1145$ |
| N ₂ | 23.446 | $23.446/28 = 0.8373$ |
| | | 1.023 |

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$$\frac{0.07165}{1.023} \times 100 = 7\%$$

$$\frac{0.1145}{1.023} \times 100 = 11.2\%$$

$$\frac{0.8373}{1.023} \times 100 = 81.8\%$$

